

Ground Water Quality AGWQMP Fact Sheets and Technical Reports

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Ohio Water Resources Council
Water Resource Monitoring Meeting
March 21, 2014



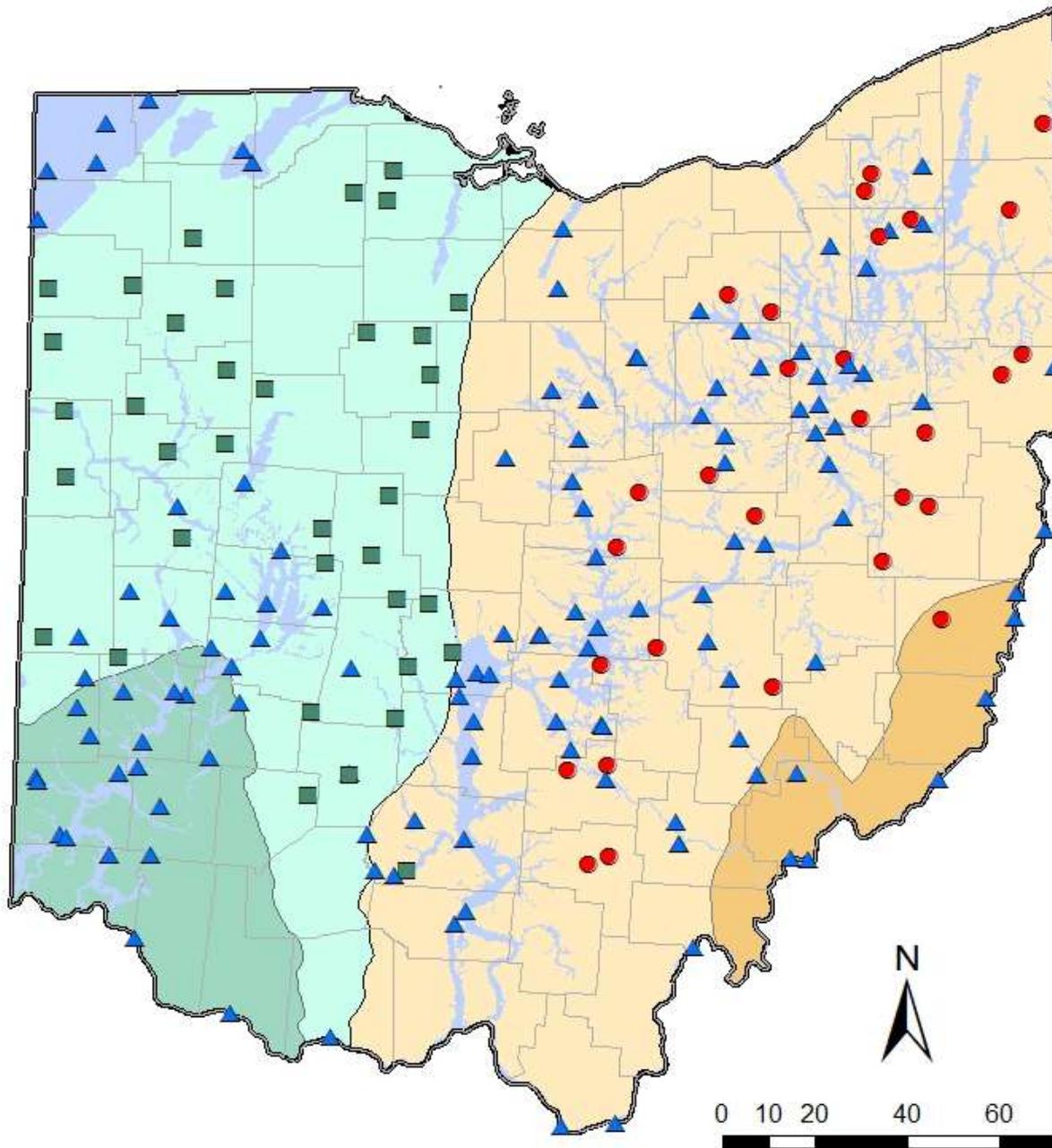
AGWQMP Well Locations and Major Aquifer Types

Active AGWQMP Wells

- ▲ Sand & Gravel Wells
- Sandstone Wells
- Carbonate Wells

Aquifer Lithology

- Sand and Gravel Aquifers
- Interbedded Sandstone/Shale
- Sandstone Aquifers
- Carbonate Aquifers
- Interbedded Carbonate/Shale



Developed using ODNR
State Aquifer Maps
March 2014

Ambient GW Quality Monitoring Program

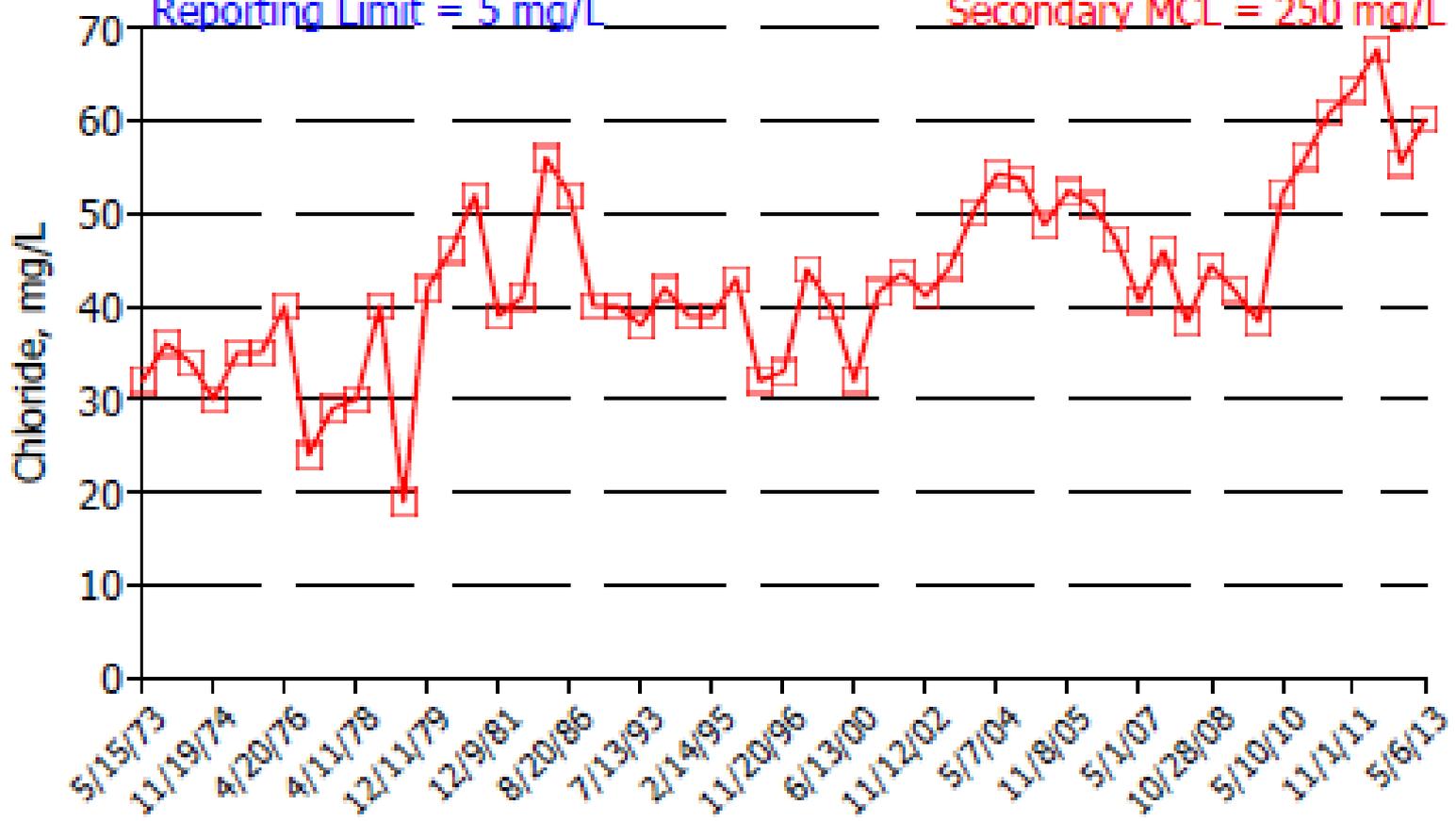
- We have a lot of water quality data;
- Using it to characterize Ohio's ground water;
- Two approaches:
 - Use the times series to identify long term water quality trends;
 - Develop Fact Sheets and Technical Reports for specific parameters.



Chloride, Total

Reporting Limit = 5 mg/L

Secondary MCL = 250 mg/L



Fact Sheets and Technical Reports

- Fluoride
- Aquifer and Ground Water Quality
 - In conjunction with “Know Your Well Water”
 - Click on “Know Your Well Water” tab on:
<http://ohiowatershed.osu.edu>
- Oxidation–Reduction
- Radionuclides

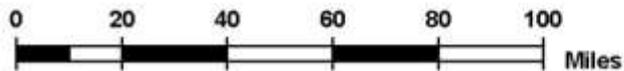
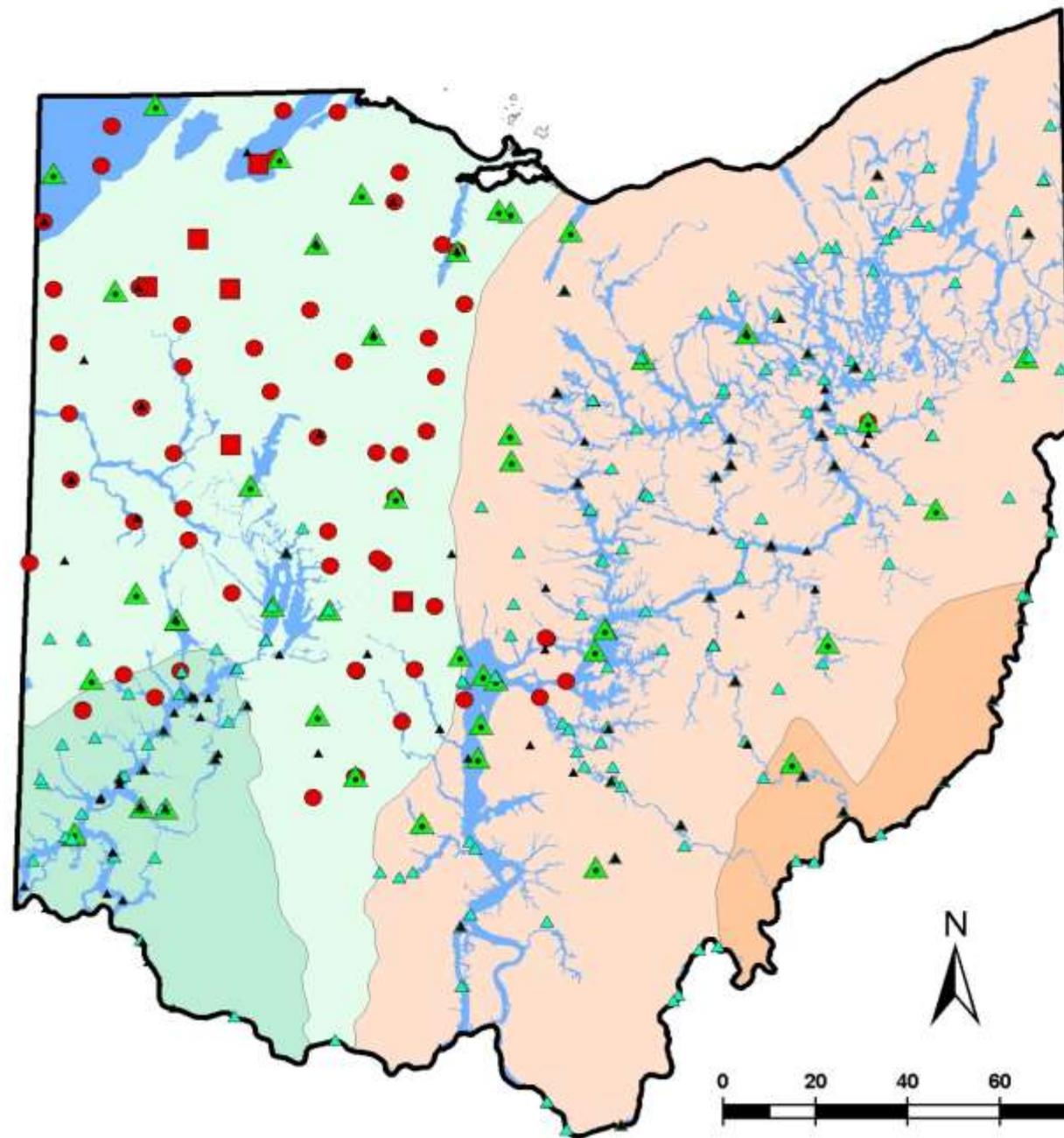


Fluoride Concentrations in Ohio Raw Water

FLUORIDE mg/L

- ▲ < 0.10
- ▲ 0.10 - 0.50
- ▲ 0.50 - 1.00
- 1.00 - 2.00
- > 2.00

-  Sand & Gravel Aquifers
-  Interbedded Shale/SS
-  Sandstone Aquifers
-  Carbonate Aquifers
-  Interbedded Shale/Carbonate



OhioEPA

Geology simplified from
ODNR Aquirer Maps
Division of Drinking and
Ground Waters
September 2010



Fluoride in Ohio's Ground Water



March 2012
Fact Sheet 2012-01
Series on Ohio's Ground Water Quality

Fluoride is the naturally-occurring stable form of the gaseous element fluorine (F). Fluoride is among the top 15 most abundant components of the Earth's crust and is naturally found in very small amounts in most aquifers. An aquifer is an underground unit of saturated earth materials that can provide usable quantities of ground water to a well.

How does fluoride get into Ohio's ground water?

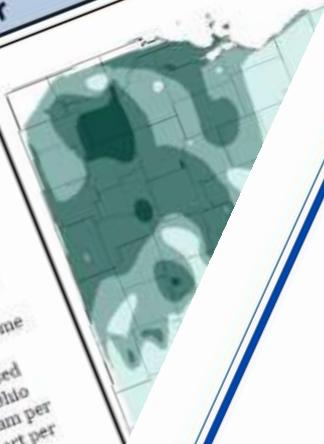
As water flows through earth materials, some of the rocks and minerals it contacts are dissolved and fluoride is naturally released into the water. Most ground waters in Ohio naturally contain less than one milligram per liter (mg/L) of fluoride (1 mg/L is 1 part per million; or one cent in a million pennies). A number of human activities can also increase fluoride concentrations in soil and water, including glass, steel, and phosphate fertilizer production. In addition, agricultural run-off, infiltration of fertilizers, and discharges from septic or sewage treatment facilities that process fluoridated water add inorganic fluorides to the environment.

Where will you find elevated fluoride?

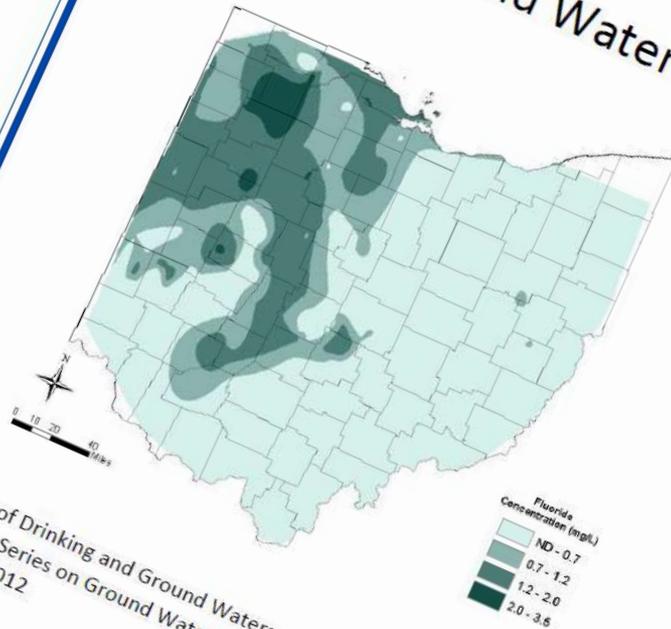
The limestone aquifers in western Ohio have the highest fluoride concentrations shown in the figure above. Limestone aquifers contain the mineral calcite, which naturally releases fluoride into ground water in the limestone. In some areas of the state, the limestone may contain some fluoride-bearing carbonate minerals. In these areas, the fluoride concentrations in some sand and gravel aquifers falls within the range of concentrations in the limestone aquifers and the lower concentrations in the sand and gravel aquifers.

What are safe levels of fluoride in drinking water?

The U.S. Environmental Protection Agency's maximum contaminant level (MCL) for fluoride in drinking water is 4 mg/L. The U.S. Environmental Protection Agency's maximum contaminant level goal (MCLG) for fluoride in drinking water is 0.7 mg/L.



Fluoride in Ohio's Ground Water



Division of Drinking and Ground Waters
Technical Series on Ground Water Quality
February 2012



FACT SHEET

Division of Drinking and Ground Waters
February 2014

Major Aquifers and Ground Water Quality in Ohio

Ohio has abundant ground water resources, but the amount available at a given location depends on local geology and well construction. Water quality also varies from place to place and over time, depending on the local geology, sensitivity and land use.

This fact sheet identifies the types of geologic materials that provide water to wells, discusses factors that influence water quality, and provides typical water quality test results.

Where does my well water come from?

Ground water is replenished by rain and melted snow infiltrating through the soil column. Average precipitation in Ohio ranges between 30 to 44 inches a year. Much of the water entering the soil is used by plants. A small amount is held in the soil, and the rest (3 to 16 inches per year) moves downward (called recharge). Any zone of saturated earth materials that is capable of providing useable quantities of water to a well is called an aquifer. The surface of the first or shallowest saturated zone or aquifer is called the water table. The amount of ground water that can flow through soils or rock depends on the size of the spaces in the earth material and how well they are connected. In most aquifers, ground water moves slowly (typically 10 to 1,000 feet per year).

Are all aquifers the same?

No. Ohio has three major aquifer types, as illustrated in Figure 1, including:

- sand and gravel deposits (blue);
- sandstone bedrock (beige); and
- carbonate bedrock (light green).

The sand and gravel aquifers consist of loose (unconsolidated) sand and gravel units with fine grained interbedded material like silts and clay or till. Most of these aquifers occur in old river valleys that have been filled with unconsolidated material and are frequently referred to as buried valley aquifers. In northwest Ohio, the unconsolidated aquifers (patches of blue) are sheets of sand and gravel deposited in glacial lake environments. Water production (yield) from the coarse-grained and thick sand and gravel aquifers range up to 500 to 1,000 gallons per minute (gpm), but are generally much lower.

The sandstone bedrock aquifers are consolidated (cemented) rock layers, interbedded (like a layer cake) with siltstones and shales. A well may receive water from multiple sandstone layers. Thicker sandstones can yield 25–100 gpm, but 25 gpm is good. Many wells in the southeast portion of Ohio produce 5 gpm and less. Where shale dominates, like the sediments (brown) in southeast Ohio, yields are low.

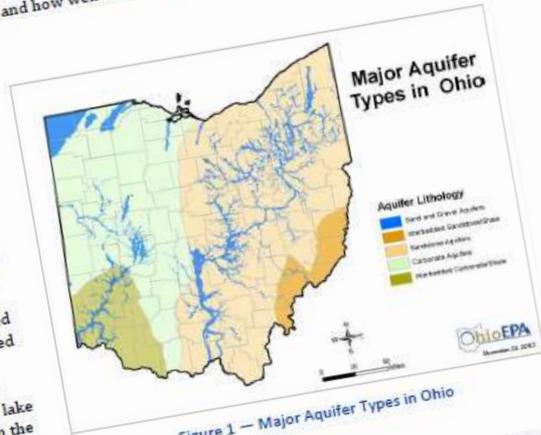


Figure 1 — Major Aquifer Types in Ohio

Major Aquifers in Ohio and Associated Water Quality

Technical Series
DRAFT

1/28/2014

Abstract

Aquifers are described and ground water quality data is presented that characterizes them. Presented provides ranges of constituent concentrations typical of the major aquifers across the state. The data are representative of source water utilized by public water systems (raw or untreated water) and are not pristine, since a number of the AGWQMP wells are impacted by elevated organic parameters sourced from surface activities. The inherent variability in water quality must be taken when extrapolating point data beyond the collection site. The data compiled in this report is the best summary available for the general water user, and is presented to help evaluate water quality in local aquifers.

Introduction

The major aquifers; water quality within aquifer types; and water quality evaluations by providing ranges of water quality across Ohio for comparison. The water quality is relative to the Ambient Ground Water Quality Monitoring Program.

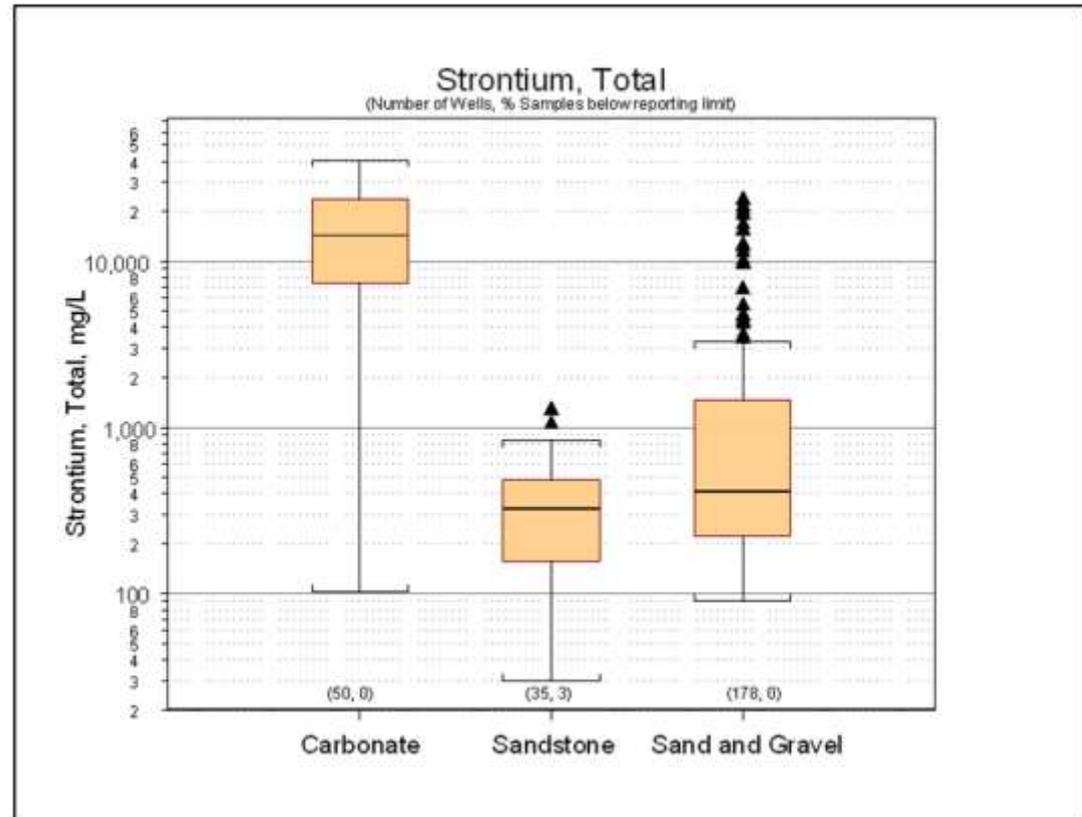
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| µg/L | µg/L | Carbonate | 71.8 | 11 | <10 | 4090 | 265 | 7 |
|------------|--|-----------------|------------|--------------|-------------------|---------------|--------------------|----------|
| NUTRIENTS | | | | | | | | |
| MCL/SMCL | Parameter and Units | Major Aquifer | Mean Value | Median Value | Minimum Value * £ | Maximum Value | Standard Deviation | Numl Sam |
| | Ammonia mg/L | Sand and Gravel | 0.22 | 0.08 | 0 | 3.41 | 0.36 | 34 |
| | | Sandstone | 0.37 | 0.17 | 0 | 2.30 | 0.49 | 7 |
| | | Carbonate | 0.41 | 0.35 | 0 | 5.93 | 0.50 | 8 |
| | Chemical Oxygen Demand mg/L | Sand and Gravel | 13.3 | <10 | <2.0 | 200 | 9.25 | 33 |
| | | Sandstone | 13.4 | <10 | <6.0 | 172 | 8.43 | 7 |
| | | Carbonate | 14.0 | <10 | <10 | 371 | 16.1 | 8 |
| 10 mg/L | Nitrite & Nitrate NO2 +NO3 as N mg/L | Sand and Gravel | 0.72 | 0.40 | 0 | 4.00 | 1.07 | 33 |
| | | Sandstone | 0.72 | 0.40 | 0 | 4.00 | 1.07 | 7 |
| | Phosphorus mg/L | Sandstone | 0.07 | 0.03 | 0 | 0.30 | 0.07 | 8 |
| | Total Kjeldahl N mg/L | Sandstone | 0.07 | 0.03 | 0 | 0.30 | 0.07 | 31 |
| | | Sand and Gravel | 0.07 | 0.03 | 0 | 0.30 | 0.07 | 6 |



Fact Sheets and Technical Reports

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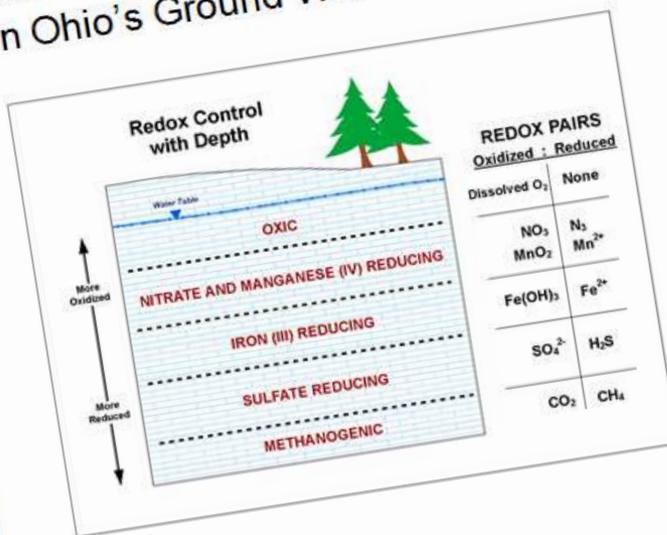


January 2014



Division of Drinking and Ground Waters
Technical Series on Ground Water Quality

Reduction-Oxidation (Redox) Control in Ohio's Ground Water Quality



John R. Kasich, Governor
Mary Taylor, Lt. Governor
Craig W. Butler, Director

DRAFT



October 2013
 Fact Sheet 201301
 Series on Ohio's Ground Water Quality

Radionuclides in Ohio's Ground Water

What are radionuclides?

Radionuclides are elements with unstable atoms. When these unstable atoms decay, they release energy (radiation) in alpha or beta particles and gamma rays. Energy is continually released until a stable, non-radioactive substance is formed. Most radionuclides occur naturally, but some come from man-made sources. The most common radionuclides found in ground water are decay products of uranium and thorium and include radium and radon.

How do radionuclides get into Ohio's ground water?

Most of the radionuclides found in Ohio's ground water occur naturally from the weathering and dissolution of rocks and minerals. The amount and type of radiation released during the decay process depends on the radionuclides present. Radioactive materials are also used in medical diagnostics and treatments, electricity production, commercial products, research and nuclear weapons. Human activities, such as disposal of radioactive wastes, may increase the levels of radioactive materials found locally in ground water.

How are radionuclides measured?

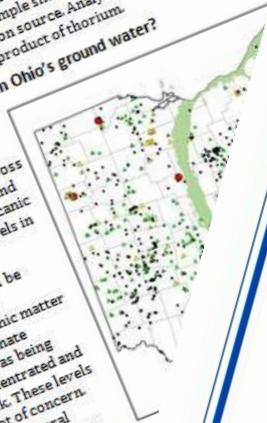
Gross alpha and gross beta particle activity are measures of the total radioactivity produced during decay and are measured in picocuries per liter (pCi/L). If a sample shows high gross alpha or gross beta, additional analysis is required to identify the specific radiation source. Analysis is also completed for individual radionuclides like radium-226, which is a decay product of thorium.

Where will you find elevated radionuclides in Ohio's ground water?

Ohio's public water systems are required to sample for radionuclides in ground water. The figure on the right shows that gross alpha in ground water is generally low across Ohio. This is expected because uranium and thorium, typically found in igneous (volcanic or plutonic) rocks, are found in low levels in Ohio's major aquifers.

The highest levels of gross alpha can be found in the carbonate bedrock of northwestern Ohio. Because organic matter was more abundant in the carbonate sediments when the bedrock was being formed, the uranium was concentrated and incorporated into the bedrock. These levels are still relatively low and not of concern. Radium-226 follows the same general pattern as gross alpha because its source material is also uranium.

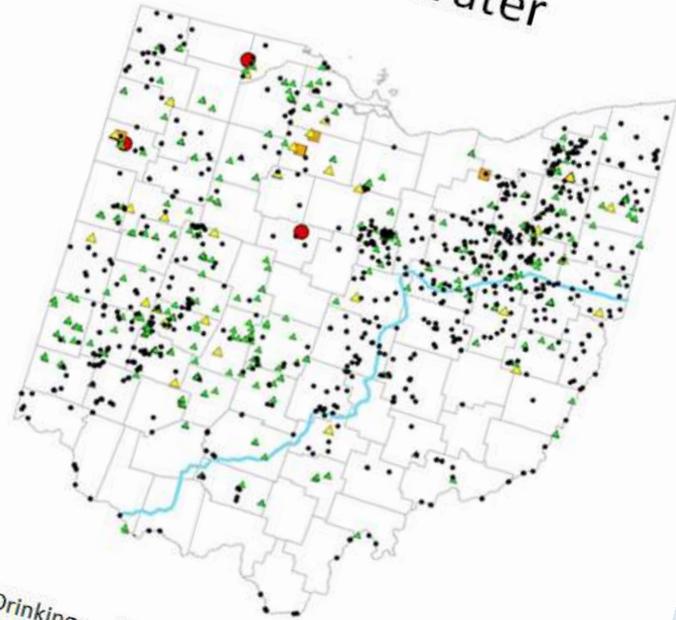
Black shales, such as the Ohio and Glenora shales, yield much water. Consequently, little public water



Ohio Department of Health
www.ohio.gov



Radionuclides in Ohio's Ground Water



Division of Drinking and Ground Waters
 Technical Series on Ground Water Quality
 October 2013

Monitoring Data – Needs to be Used

- Value of data is documented by its use;
- Use of data helps inform the public;
- Provides the basis to help make policy decisions and individual decisions;

Radionuclide Fact Sheet and Technical Report under review in Ohio EPA Directors Office?



Acknowledgements

- Linda Slattery, Mike Slattery, and Jeff Patzke have prepared/helped prepare documents;
- Ohio EPA district staff collect AGWQMP samples;
- PWS operators – Their cooperation is critical for AGWQMP efforts.



Ohio EPA

Division of Drinking and Ground Waters

Ground Water Quality Characterization Program

<http://www.epa.state.oh.us/ddagw/gwqcp.aspx>

